

A SYNTHETIC APPROACH TO NEW 1,4-BENZODIAZEPINE DERIVATIVES

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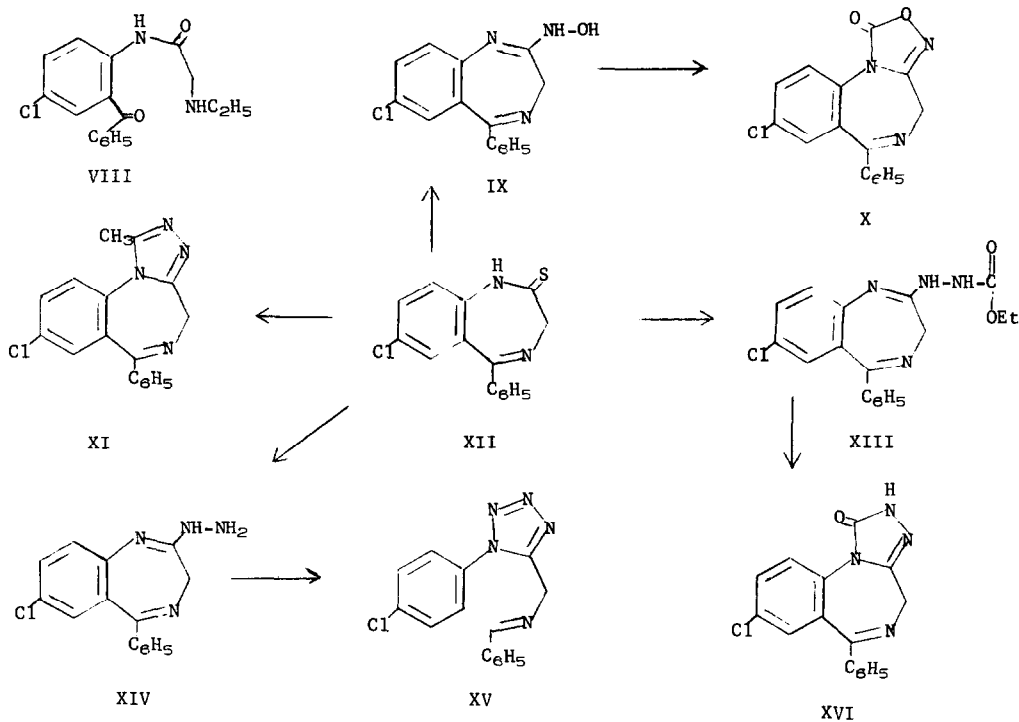
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A recent publication by Meguro and Kuwada¹ prompts us to report some of our results in this area. In particular we have studied the utility of electrophilic amide derivatives for the preparation of 1,4-benzodiazepines with heterocyclic systems fused to the "a" and the "d" faces of the benzodiazepine system.

An electrophilic center at C-5 was created by alkylation of 7-chloro-1,2,3,4-tetrahydro-1-methyl-5H-1,4-benzodiazepin-5-one^{2,3}, mp 183-185°, with triethyloxonium fluoroborate⁴ to give 7-chloro-5-ethoxy-2,3-dihydro-1H-1,4-benzodiazepine (II), mp 38.5-39°. This compound (II) readily condensed with amines to give amidines. Thus reaction of II with N,N-diethylethylenediamine in refluxing benzene gave 7-chloro-5-[[2-(diethylamino)ethyl]amino]-2,3-dihydro-1-methyl-1H-1,4-benzodiazepine (I) which was isolated as its dihydrochloride salt, mp 253-255°.

Addition of hydroxyl amine to II in refluxing ethanol gave 7-chloro-2,3-dihydro-5-(hydroxyamino)-1-methyl-1H-1,4-benzodiazepine (V), mp 186.5-189°; which was condensed with carbonyl-diimidazole in refluxing THF to give 10-chloro-6,7-dihydro-7-methyl-3H,5H-[1,2,4]oxadiazolo[4,3-d][1,4]benzodiazepin-3-one (VI), mp 190-191°. The reaction of II with acid hydrazides proceeded with concomitant cyclodehydration to give s-triazolo[4,3-d][1,4]benzodiazepines. Thus condensation of II with acetic acid and benzoic acid hydrazides gave 10-chloro-6,7-dihydro-3,7-dimethyl-5H-s-triazolo[4,3-d][1,4]benzodiazepine (III), mp 185-186.5°, and 10-chloro-6,7-dihydro-7-methyl-3-phenyl-5H-s-triazolo[4,3-d][1,4]benzodiazepine (IV), mp 207.5-208.5°, respectively. Similarly the reaction of II with ethyl carbazate gave 10-chloro-2,5,6,7-tetrahydro-7-methyl-3H-s-triazolo[4,3-d][1,4]benzodiazepin-3-one (VII), mp 259-262°.

Chart 2



A sample of the monohydrobromide salt of XI was dissolved in 95% ethanol and allowed to crystallize slowly at ambient temperature. The crystals were shown by x-ray crystallographic analysis to be the hemihydrobromide ethanol solvate. The crystals are triclinic, space group $P\bar{1}$, with cell dimensions $a=7.54\text{\AA}$, $b=14.62\text{\AA}$, $c=17.61\text{\AA}$, $\alpha=94.2^\circ$, $\beta=92.5^\circ$, $\gamma=100.9^\circ$. The symmetry independent unit contains two molecules of XI, two ethanols, and one HBr. Final results of a three-dimensional x-ray analysis ($R=0.059$) show that one of the molecules is protonated at N-2 of the triazole ring. Both the protonated and the unprotonated molecules have the same conformation (shown below for the protonated molecule). The protonated triazole nitrogen forms a strong hydrogen bond ($N\cdots O$ distance of 2.66\AA) to one of the ethanol oxygens. The ethanols are hydrogen-bonded to the Br ion.

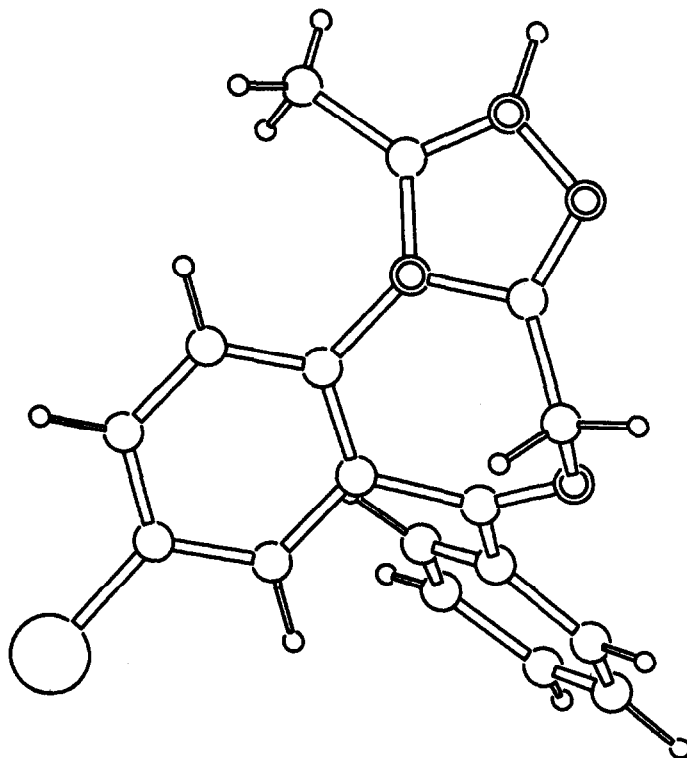


Figure. Computer drawing XI from x-ray results for the protonated molecule (see text).

REFERENCES

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6. G.A. Archer and L.H. Sternbach, J. Org. Chem., 29, 231 (1964).
7. The authors wish to acknowledge the able technical assistance of Mr. J. Robert Greene.
8. A complete report of this work will be published elsewhere.